



Logarithms

Logarithms

Logarithms are useful

- when quantities vary on many orders of magnitude
 - $[H^+] \rightarrow pH = -\log [H^+]$
- Solving some problems involving exponents

Logarithms

Logarithms and exponentiation are inverse functions

- one will undo the other

If $y = b^x$

Then the logarithm of y to base b is x

$$\log_b(y) = x$$

The logarithm of y is the number, x, to which the base, b, must be raised to equal y

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Logarithms

In chemistry, the only bases used are base 10 and base e

- Base 10 logarithms
 - common logarithm
 - written as log, the b is omitted
 - “log” key on your calculator
- Base e logarithms
 - natural logarithms
 - $e = 2.718 \dots$
 - written as ln
 - “ln” key on your calculator

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Logarithms

Try these

	log	ln
734		
0.0150		
4.47×10^{-5}		

Then use the exponential function to get the original number back.

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Logarithms

Try these

	log	ln
734	2.866	6.599
0.0150	-1.824	-4.200
4.47×10^{-5}	-4.350	-10.016

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Logarithms

Useful relationships valid for either log or ln

- $\log(xy) = \log(x) + \log(y)$
 $x = 10^{\log(x)}$ and $y = 10^{\log(y)}$
 $xy = 10^{\log(x)} \times 10^{\log(y)} = 10^{\log(x) + \log(y)}$
 $\therefore \log(xy) = \log(x) + \log(y)$
- $\log(x/y) = \log(x) - \log(y)$
- $\log(x^a) = a \log(x)$
 $\log(x^2) = \log(x \times x) = \log(x) + \log(x) = 2\log(x)$

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Logarithms

To convert between log and ln

- $\ln(x) = \ln(10) \log(x) = 2.303 \log(x)$
 $x = e^{\ln(x)} = 10^{\log(x)} = (e^{\ln(10)})^{\log(x)} = e^{\ln(10)\log(x)}$
 $\therefore \ln(x) = \ln(10) \log(x)$

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Logarithms

In lab 2 logarithms will be used to find the order of a reaction
The rate of reaction will be measured for different reactant concentrations.

Will have to solve equations such as

$$0.464 = 0.500^x$$

Take log of both sides

$$\log(0.464) = \log(0.500^x) = x \log(0.500)$$

$$x = \frac{\log(0.464)}{\log(0.500)} = 1.11$$

Try solving the following equation for x

$$2.3 = (1.5)^x$$