

CHEM 0010 Tutorial Answers

A. Mole Questions 1

- 10.0 mole Li = 69.0 g Li
- 3.00 mole of Mg = 1.81×10^{24} atoms Mg
- 5.00×10^{23} atoms C = 0.830 mole C
- 7.25 g Cu = 6.88×10^{22} atoms Cu
- 2.22×10^{26} atoms Pb = 3.69×10^2 mole Pb
- 15.0 mole K = 9.03×10^{24} atoms K
- 4.65×10^{22} atoms S = 0.0772 mole S
- 333 g Mg = 13.7 mole Mg
- 48.0×10^{23} atoms O = 128 grams O
- 0.250 mole Br = 1.5×10^{23} atoms Br
11. Define mole.

Mole is a counting unit, a quantity of things (1 mole of any substance contains 6.022×10^{23} particles)

12. Define molar mass.

Molar mass is a mass (in grams) of one mole of that substance. (Example: Molar mass of oxygen atom is 16 g/mol)

13. State Avogadro's number.

$$6.022 \times 10^{23}$$

Mole Questions 2

- 1.77×10^{28} atoms Cu = 2.94×10^4 mole Cu = 1.87×10^6 grams Cu
- 44.0 g Ca = 1.10 mole Ca = 6.62×10^{23} atoms Ca
- 0.775 mole K = 4.67×10^{23} atoms K = 30.3 g K
- 1.99×10^{25} molecules of H₂O = 33.0 moles H₂O
- 0.667 moles Li = 4.60 g Li
- 5.88 moles Na = 3.54×10^{24} atoms Na
- 15.0×10^{23} atoms Zn = 163 g Zn
- 878 g O = 3.30×10^{25} atoms O
- 125 g Cr = 1.45×10^{24} atoms Cr
- 4.75×10^6 atoms Al = 2.13×10^{-16} g Al

Mole Questions 3

1. 5.00 mole Zn = 3.01×10^{24} atoms Zn
2. 2.00×10^{47} molecules H_2O = 3.32×10^{23} mole H_2O
3. 1.50 mole Ca = 60.2 grams Ca
4. 6.75 mole Cu = 4.06×10^{24} atoms Cu
5. 1.75×10^{15} atoms K = 2.91×10^{-9} mole K
6. 5.00g C = 2.51×10^{23} atoms C
7. 45.9 g Pb = 0.222 mole Pb
8. 3.01×10^{23} molecules CO_2 = 0.5 mole CO_2
9. 4.25×10^{22} atoms Ni = 4.14 g Ni
10. 9.50 g Li = 8.29×10^{23} atoms Li
11. 0.250 mole Si = 1.51×10^{23} atoms Si
12. 2.30 mole Mg = 55.9 g Mg
13. 2.50×10^{25} atoms Au = 41.5 mole Au
14. 105 g Be = 11.7 mole Be
15. 75.0 g S = 1.41×10^{24} atoms S
16. Which contains a greater number of atoms, 2.0 moles of Zn or 2.0 moles Cu?
They both contain $2 \times 6.022 \times 10^{23} = 1.204 \times 10^{24}$
17. Which has a greater mass, 2.0 moles of Zn or 2.0 moles Cu? Zn
18. A mole of moles digs a mole of holes. How many holes does each mole dig?
One.